

for apphu 10/507,399



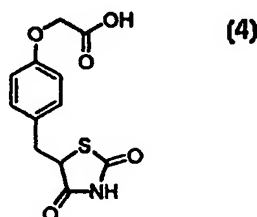
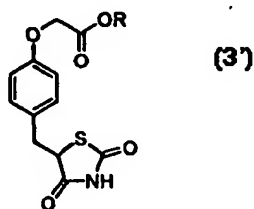
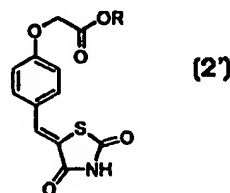
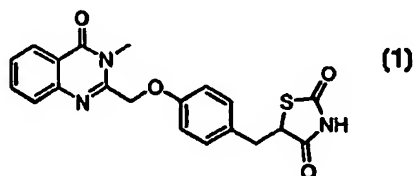
PCT

WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau

## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification 7 : <b>C07D 417/12, 277/24</b>		A1	(11) International Publication Number: <b>WO 00/15638</b>
			(43) International Publication Date: 23 March 2000 (23.03.00)
(21) International Application Number: <b>PCT/IB99/01530</b>		(74) Common Representative: <b>DR. REDDY'S RESEARCH FOUNDATION; 7-1-27, Amcerpet, Hyderabad 500 016 (IN).</b>	
(22) International Filing Date: 10 September 1999 (10.09.99)			
(30) Priority Data: 2060/MAS/98 14 September 1998 (14.09.98) IN		(81) Designated States: AE, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, UA, UG, US, UZ, VN, YU, ZA, ZW, ARIPO patent (GH, GM, KE, LS, MW, SD, SL, SZ, UG, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG).	
(71) Applicant (for all designated States except US): <b>DR. REDDY'S RESEARCH FOUNDATION [IN/IN]; 7-1-27, Amcerpet, Hyderabad 500 016 (IN).</b>			
(72) Inventors; and			
(75) Inventors/Applicants (for US only): <b>CHEBIYYAM, Prabhakar [IN/IN]; Dr. Reddy's Research Foundation, 7-1-27, Amcerpet, Hyderabad 500 016 (IN). POTLAPALLY, Rajender, Kumar [IN/IN]; Dr. Reddy's Research Foundation, 7-1-27, Amcerpet, Hyderabad 500 016 (IN). GADE, Chinna, Bakki, Reddy [IN/IN]; Dr. Reddy's Research Foundation, 7-1-27, Amcerpet, Hyderabad 500 016 (IN). SATISH, Balaram, Mahanti [IN/IN]; Dr. Reddy's Research Foundation, 7-1-27, Amcerpet, Hyderabad 500 016 (IN). MAMILLAPALLI, Ramabhadra, Sarma [IN/IN]; Dr. Reddy's Research Foundation, 7-1-27, Amcerpet, Hyderabad 500 016 (IN). GAD-DAM, Om, Reddy [IN/IN]; Dr. Reddy's Research Foundation, 7-1-27, Amcerpet, Hyderabad 500 016 (IN).</b>		Published With international search report.	

(54) Title: AN IMPROVED PROCESS FOR THE PREPARATION OF THIAZOLIDINE-2,4-DIONE DERIVATIVES



## (57) Abstract

An improved process for the preparation of 5-[4-[[3-Methyl- 4-oxo-3,4- dihydroquinazolin -2-yl]methoxy] benzyl] thiazolidine -2,4-dione of formula (1) which comprises: reducing the compound of formula (2') where R represents a (C<sub>1</sub>-C<sub>4</sub>)alkyl group using Raney Nickel or Magnesium in alcohol having 1 to 4 carbon atoms or mixtures thereof, if desired reesterifying using sulphuric acid at a temperature in the range of 0 °C to 60 °C to obtain a compound of formula (3') wherein R is as defined above, hydrolyzing the compound of formula (3') wherein R is as defined above, by conventional methods to obtain the acid of formula (4), condensing the acid of formula (4) with N-methyl anthranilamide directly without any preactivation of the acid to produce the compound of formula (1) and if desired, converting the compound of formula (1) to pharmaceutically acceptable salts thereof by conventional methods.

**FOR THE PURPOSES OF INFORMATION ONLY**

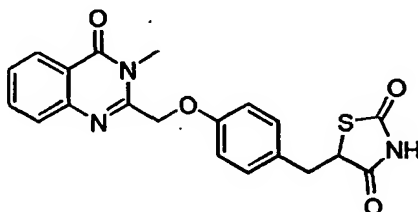
Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav Republic of Macedonia	TM	Turkmenistan
BF	Burkina Faso	GR	Greece	ML	Mali	TR	Turkey
BG	Bulgaria	HU	Hungary	MN	Mongolia	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MR	Mauritania	UA	Ukraine
BR	Brazil	IL	Israel	MW	Malawi	UG	Uganda
BY	Belarus	IS	Iceland	MX	Mexico	US	United States of America
CA	Canada	IT	Italy	NE	Niger	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NL	Netherlands	VN	Viet Nam
CG	Congo	KE	Kenya	NO	Norway	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NZ	New Zealand	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's Republic of Korea	PL	Poland		
CM	Cameroon	KR	Republic of Korea	PT	Portugal		
CN	China	KZ	Kazakhstan	RO	Romania		
CU	Cuba	LC	Saint Lucia	RU	Russian Federation		
CZ	Czech Republic	LI	Liechtenstein	SD	Sudan		
DE	Germany	LK	Sri Lanka	SE	Sweden		
DK	Denmark	LR	Liberia	SG	Singapore		
EE	Estonia						

## **An Improved Process for the Preparation of thiazolidine-2,4-dione derivatives**

### **Field of Invention**

The present invention relates to an improved process for the preparation of thiazolidine-2,4-dione derivatives. More particularly the present invention relates to an improved process for the preparation of 5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1) and pharmaceutically acceptable salts thereof which are useful as antidiabetic compounds. The thiazolidine-2,4-dione derivative of the formula (1) is particularly useful for the treatment of diabetes type II (NIDDM) and related complications.



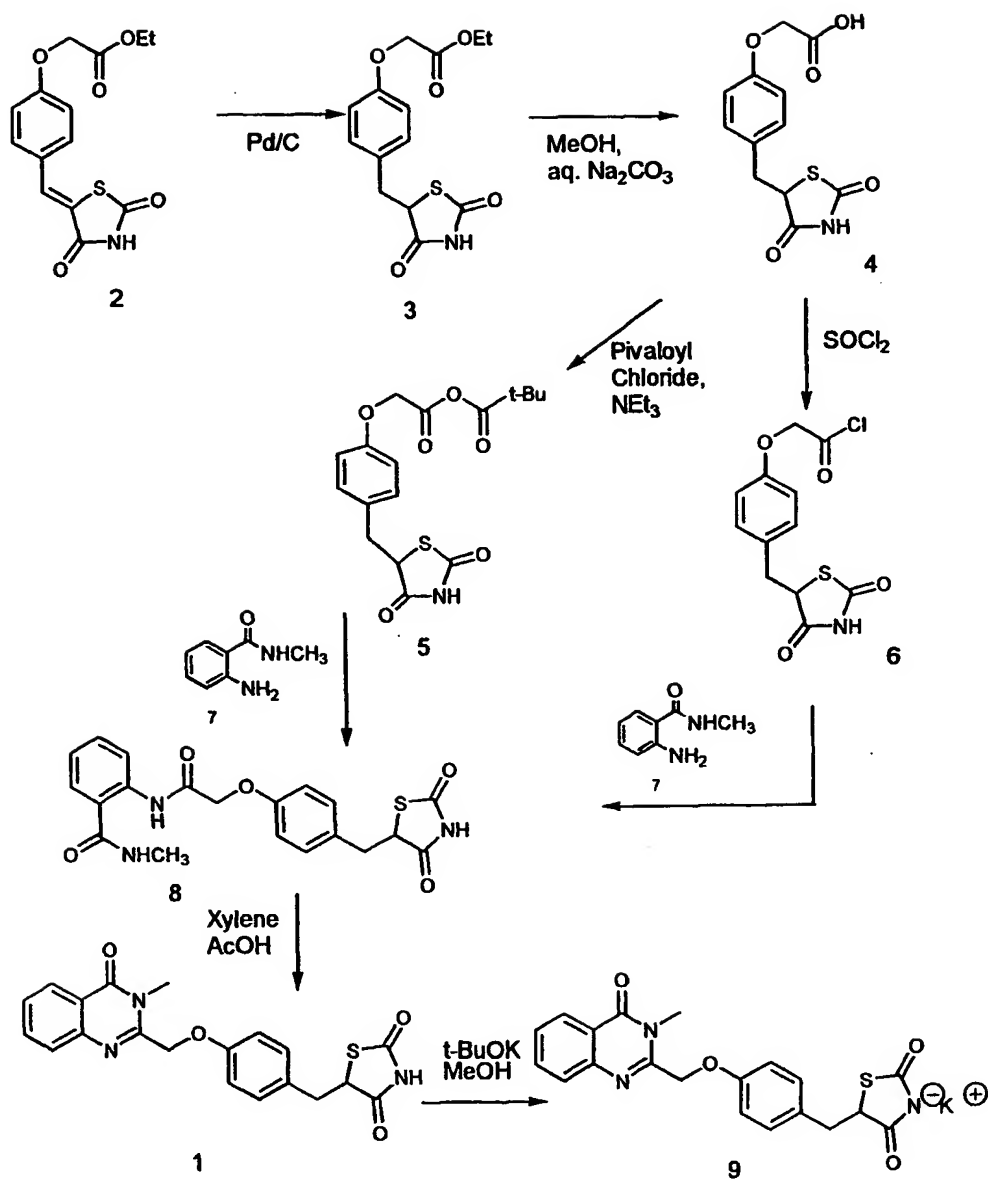
(1)

### **Background of the invention**

We have in our international publication number WO 97/41097 described the synthesis of the 5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1). Compound of the formula (2) on reduction using the expensive catalyst Pd/C in stoichiometric quantity gives the corresponding saturated compound of the formula (3). The ethyl ester of the formula (3) on hydrolysis using methanol/water/sodium carbonate recipe gives the acid of the formula (4) in about 80% yield after a tedious workup sequence involving removal of methanol, then dilution with

water, extraction with an organic solvent to remove impurities and then adjustment of pH to precipitate the required acid of the formula (4). The acid of the formula (4) is activated by converting it either to the mixed anhydride of the formula (5) by treating with pivaloyl chloride or the acid chloride of the formula (6) by treating with thionylchloride. Condensation of formula (5) or (6) with N-methyl anthranilamide of the formula (7) gives the amide of the formula (8). Amide of the formula (8) on cyclisation by refluxing in xylene/acetic acid for ~20-30 hours yields about 50% of the cyclised compound of the formula (1). Compound of the formula (1) upon treatment with potassium t-butoxide in methanol gives the corresponding potassium salt of the formula (9). The reaction steps involved in the process are shown in scheme-I below.

3



Scheme-I

The following are the difficulties encountered during the scaleup trials employing the above said process:

- The step of preparing the compound of the formula (3) requires stoichiometric quantities of Pd/C. Nearly 70 % of the total cost of the product is due to the use of Pd/C which is very expensive. The time required for the completion of the reaction is about 40 hours, which is also very high and further escalates the cost.
- The hydrolysis of compound of formula (3) to give the acid of the formula (4) by using methanol/water/sodium carbonate recipe makes the reaction workup more tedious because it involves removal of methanol, then dilution with water, extraction with an organic solvent to remove impurities and then adjustment of pH to precipitate the required acid of the formula (4). In addition, the reaction time is large, i.e. more than 12 hours. Further the yield is also not very good (80 %).
- The activation of the acid of the formula (4) by converting to the mixed anhydride of the formula (5) involves use of different chemicals such as pivaloyl chloride, triethylamine and solvents such as dichloromethane, which results in messing-up of the reaction mixture. Further more the conversion of the acid of the formula (4) to the acid chloride of the formula (6) involves the use of corrosive reagents like thionyl chloride. Moreover, the reactions are moisture sensitive.
- Because of the large number of chemicals employed in the previous step, the isolation of the intermediate amide of the formula (8) becomes very complicated and also results in low yield (50 %) of the amide of the formula (8).

- The cyclisation of the intermediate amide of the formula (8) results in low yield (~50%) of the compound of the formula (1) and the reaction time is large (~40 hours).
- The preparation of potassium salt of the formula (9) employing potassium t-butoxide is not only risky but also expensive thereby making the process uneconomical.

Keeping in view of the above difficulties in the process disclosed in our copending application mentioned above for the preparation of 5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1), we directed our research towards developing an improved process which would be cost and time effective, as well as simple for scaling-up.

### **Objective of the Invention**

The main objective of the present invention is, therefore, to provide an improved process for the preparation of 5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1) avoiding the above mentioned difficulties.

Another objective of the present invention is to provide an improved process for the preparation of 5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1) without employing expensive and hazardous chemicals thereby making the process not only economical but also safe.

Yet another objective of the present invention is to provide an improved process for the preparation of 5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1), which involves very simple work-up procedures making the process simple.

We have developed the improved process of the present invention based on our finding that use of Raney-Nickel or magnesium/ methanol as reducing agents to reduce the compound of the formula (2') where R represents a (C<sub>1</sub>-C<sub>4</sub>)alkyl group, not only results in the reduction of cost but also results in efficient reduction. In addition, the compound of formula (3') where R represents a (C<sub>1</sub>-C<sub>4</sub>)alkyl group and the compound of formula (4) can also be directly condensed with N-methyl anthranilamide of the formula (7) without preactivation to produce compound of formula (1) which further makes the process simple and economical.

#### **Detailed description of the invention:**

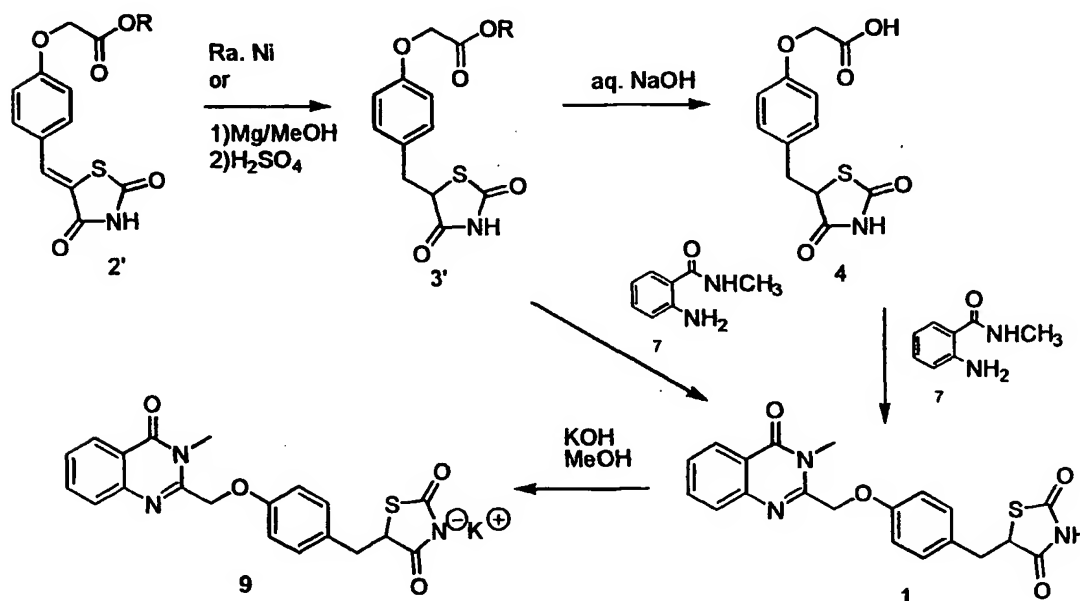
Accordingly the present invention provides an improved process for the preparation of 5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1), which comprises :

- (a) reducing the compound of the formula (2') where R represents a (C<sub>1</sub>-C<sub>4</sub>)alkyl group using Raney Nickel or Magnesium in alcohol having 1 to 4 carbon atoms or mixtures thereof, and if desired reesterifying using sulphuric acid at a temperature in the range of 0 °C to 60 °C to obtain a compound of formula (3') wherein R is as defined above,
- (b) hydrolysing the compound of formula (3') wherein R is as defined above, by conventional methods to obtain the acid of the formula (4),



- (c) condensing the acid of the formula (4) with N-methyl anthranilamide of the formula (7) directly without any preactivation of the acid to produce the compound of formula (1) and if desired
- (d) converting the compound of formula (1) to pharmaceutically acceptable salts thereof by conventional methods.

According to an embodiment of the present invention, the compound of the formula (3') wherein R is as defined above, obtained in step (a) may also be condensed directly with N-methyl anthranilamide of the formula (7) to obtain the compound of the formula (1). The reaction is shown in Scheme-II below :



Scheme-II

The reduction of the compound of formula (2') wherein R is as defined above using 40-130% (w/v) preferably 100% (w/v) Raney Nickel proceeds to

completion in 8 to 70 hours, preferably from 12-24 hours, at 15 °C –70 °C preferably 30 °C –60 °C and at atmospheric pressure to 600 psi, preferably from atmospheric pressure to 400 psi of hydrogen pressure. The crude material is taken in lower alcohol like methanol, ethanol, propanol and the like and precipitated by adding water thereby affording a highly pure compound of the formula (3'), in about 85-90 % overall yield and a purity of about 97-99 %. The reduction using magnesium, (4-12 eq., preferably 8-10 eq.) in alcohol having 1 to 4 carbon atoms or their mixtures at a temperature in the range of 10 °C to 60 °C, preferably at a temperature in the range of 15 °C to 30 °C for about 2-15 hours, preferably from 6-8 hours results in a mixture of the acid of the formula (4) and an ester of the formula (3') where R is as defined above.

After reacting with magnesium/alcohol having 1 to 4 carbon atoms for 2-15 hours, preferably from 6-8 hours, either water is added and reaction continued to obtain pure compound of formula (4) or sulphuric acid is added till pH is 2 and refluxed for 2-15 hours, preferably for 6-8 hours to produce pure ester of the formula (3') where R is as defined above. The inorganic salts precipitate out quantitatively in the form of magnesium sulphate. Hence, no dissolved solids get into the effluent. These esters of the formula (3') upon hydrolysis with aq. sodium hydroxide give the acid of the formula (4) in 97-99 % yield and 95-99 % purity. The reaction time is drastically reduced to only ~2 hours as compared to 12 hours required by the process disclosed in our above said international publication. Workup is also extremely simplified involving only pH adjustment to obtain the required acid of the formula (4). The acid of the formula (4) is condensed with N-methyl anthranilamide of the formula (7) directly for about 6-20 hours, preferably 10-12 hours to produce the compound of formula (1) without

any pre-activation of the acid of the formula (4). The yield is ~70 % with a purity of ~99 %.

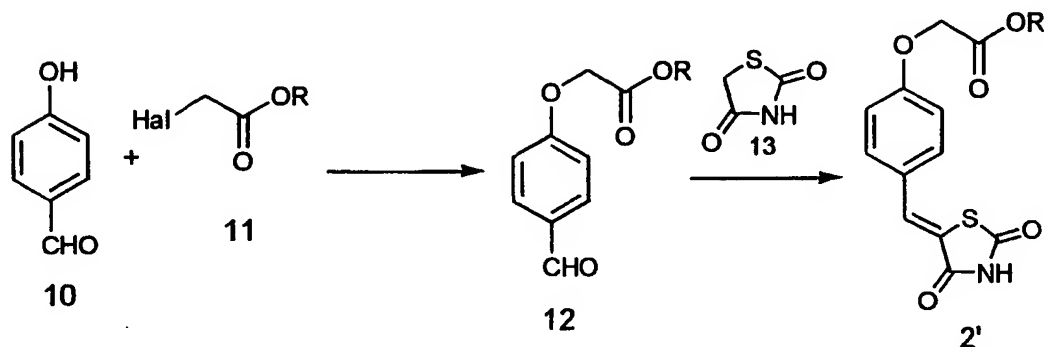
Alternatively, the condensation can also be carried out with the esters of formula (3') where R is as defined above with N-methyl anthranilamide of the formula (7) for a period of 5-30 hours, preferably 6-20 hours to produce the compound of formula (1) albeit in low yield (20 %). However, the yield can be improved to a maximum of 60 % if the reaction time is increased to 40-50 hours. The resulting compound of formula (1) upon treating with methanolic potassium hydroxide, potassium carbonate or potassium t-butoxide, at 60-70 °C and cooling the reaction mixture to room temperature and maintaining it for 1h room temperature, gives the corresponding potassium salt of the formula (9) in ~90 % yield in a pharmaceutically acceptable quality. The reaction may be carried out in the presence of solvents such as xylene/methanol mixture in the ration of 1 : 1. In a similar manner, other pharmaceutically acceptable salts of the formula (1) can be prepared by conventional methods.

The present invention also envisages an improved process for the preparation of compound of the formula (2) starting from p-hydroxybenzaldehyde of the formula (10) and alkylhaloacetate of the formula (11). This process comprises

- a). reacting p-hydroxybenzaldehyde of the formula (10) and alkylhaloacetate of the formula (11) where Hal represents halogen atom like fluorine, chlorine, bromine or iodine and R is as defined earlier in the presence of aromatic hydrocarbon solvents, a base, alkyl or aryl sulphonic acid and iodine to obtain the compound of the formula (12) where R is as defined earlier.

b). condensing the compound of formula (12) where R is as defined earlier with thiazolidine-2,4-dione of the formula (13) in the presence or absence of a solvent using catalysts to produce a compound of formula (2').

The reaction is shown in Scheme-III below :



**Scheme - III**

The reaction may be carried out in the presence of aromatic hydrocarbon solvent such as benzene, toluene, xylene and the like or mixtures thereof. The base such as alkali and alkaline earth metal carbonates and bicarbonates like potassium carbonate, potassium bicarbonate, sodium carbonate, calcium carbonate and the like may be used. The alkyl or aryl sulphonic acid such as methane sulphonic acid, ethane sulphonic acid, propane sulphonic acid, p-toluene sulphonic acid, benzene sulphonic acid, p-nitro benzene sulphonic acid and the like may be used.

We have observed that the use of iodine activates the halo group present in the compound of formula (11) where Hal represents halogen atom like fluorine, chlorine, bromine and R is as defined earlier while reflux using a Dean-Stark condenser in the presence of alkyl or aryl sulphonic acids helps in enhancing the reaction rate. The reaction is complete in 3-10 hours, preferably 5-7 hours under

these conditions as compared to ~18 hours as described in the prior art. Moreover, the reaction workup is simplified by addition of water to the reaction mixture followed by separation of solvent layer. The solvent layer is used as such for the next step of condensing the compound of formula (12) where R is as defined earlier with thiazolidine-2,4-dione of the formula (13). Since, water is being removed azeotropically, in this step, no drying of the solvent layer is required. This process not only uses a single, safe solvent but also optionally makes the two-stage process of the preparation of the compound of formula (12) where R is as defined earlier in a single pot operation. The yield and purity of the compound of the formula (12) is also found to be good (80 % and 90 % respectively).

The condensation of compound of formula (12) with compound of formula (13) may be carried out in the presence or absence of solvents such as toluene, xylene and the like, using catalysts such as benzoic acid, piperidine and the like, at reflux temperature for a period of 6-8 h to give compound of formula (2') in ~ 85 % yield.

The present invention is described in detail with examples given below which are provided by way of illustration only and therefore should not be construed to limit the scope of the invention.

#### **Example-1**

##### **Preparation of 4-((carboethoxy)methoxy)benzaldehyde**

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), toluene (2.5 L), p-toluene sulphonic acid (39 g, 0.21 M) and iodine (2 g, catalytic)

were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Ethylbromoacetate (341 g, 2.05 M) was added and the reaction refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with toluene (2 X 500 ml). The combined organic layers were washed with brine and concentrated under vacuum to yield 4-((carboethoxy)methoxy)benzaldehyde as an oily material (407 g, Y=96%, P=99%).

### **Example-2**

#### **Alternative preparation of 4-((carboethoxy)methoxy)benzaldehyde**

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), toluene (2.5 L), p-toluenesulfonic acid (39 g, 0.21 M) and iodine (2 g, catalytic) were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Ethylchloroacetate (251 g, 2.05 M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with toluene (2 X 500 ml). The combined organic layers were washed with brine and concentrated under vacuum to yield 4-((carboethoxy)methoxy)benzaldehyde as an oily material (395 g, Y=93%, P=99%).

### **Example-3**

#### **Alternative preparation of 4-((carboethoxy)methoxy)benzaldehyde**

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), xylene (2.5 L), p-toluenesulfonic acid (39 g, 0.21 M) and iodine (2 g, catalytic)

were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Ethylbromoacetate (341 g, 2.05 M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with xylene (2 X 500 ml). The combined organic layers were washed with brine and concentrated under vacuum to yield 4-((carboethoxy)methoxy)benzaldehyde as an oily material (408 g, Y=97%, P=99%).

#### **Example-4**

##### **Alternative preparation of 4-((carboethoxy)methoxy)benzaldehyde**

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), toluene (2.5 L), methanesulfonic acid (20 g, 0.21 M) and iodine (2 g, catalytic) were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Ethylbromoacetate (341 g, 2.05 M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with toluene (2 X 500 ml). The combined organic layers were washed with brine and concentrated under vacuum to yield 4-((carboethoxy)methoxy)benzaldehyde as an oily material (400 g, Y=94%, P=99%).

#### **Example-5:**

##### **Alternative preparation of 4-((carboethoxy)methoxy)benzaldehyde:**

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), toluene (2.5 L), ethanesulfonic acid (23 g, 0.21 M) and iodine (2 g, catalytic) were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Ethylbromoacetate (341 g, 2.05 M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with toluene (2 X 500 ml). The combined organic layers were washed with brine and concentrated under vacuum to yield 4-((carboethoxy)methoxy)benzaldehyde as an oily material (395 g, Y=93%, P=99%).

#### **Example-6**

##### **Preparation of 4-((carbomethoxy)methoxy)benzaldehyde**

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), toluene (2.5 L), p-toluenesulfonic acid (39 g, 0.21 M) and iodine (2 g, catalytic) were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Methylbromoacetate (314 g, 2.05 M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with toluene (2 X 500 ml). The combined organic layers were washed with brine and concentrated under vacuum to yield 4-((carbomethoxy)methoxy)benzaldehyde as an oily material (385 g, Y=97%, P=99%).

#### **Example-7**

##### **Alternative preparation of 4-((carbomethoxy)methoxy)benzaldehyde**



4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), toluene (2.5 L), p-toluenesulfonic acid (39 g, 0.21 M) and iodine (2 g, catalytic) were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Methylchloroacetate (223 g, 2.05 M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with toluene (2 X 500 ml). The combined organic layers were washed with brine and concentrated under vacuum to yield 4-((carbomethoxy)methoxy)benzaldehyde as an oily material (380 g, Y=95%, P=99%).

#### Example-8

##### Preparation of 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione

4-((Carboethoxy)methoxy)benzaldehyde obtained by following a procedure described in any of Examples 1-5 (640 g, 3.08 M), thiazolidine-2,4-dione (360 g, 3.08 M), piperidine (45 ml, 0.55 M), benzoic acid (45 g, 0.37 M) and toluene (3 L) were taken in a 5 L 4-neck round bottom flask fitted with a mechanical stirrer and a Dean-Stark condenser. The reaction mixture was refluxed for 6-8 hours, while monitoring the reaction on TLC. After the completion of the reaction, the reaction mass was cooled to 10 °C and the solid thus obtained was filtered, washed with toluene (2 X 250 ml) and dried at 80 °C for 1-2 hours, to afford 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione (790 g, Y=84%, P=98%).

**Example-9****Alternative preparation of 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione**

4-((Carboethoxy)methoxy)benzaldehyde obtained by following a procedure described in any of Examples 1-5 (640 g, 3.08 M), thiazolidine-2,4-dione (360 g, 3.08 M), piperidine (45 ml, 0.55 M), benzoic acid (45 g, 0.37 M) and xylene (3 L) were taken in a 5 L 4 neck round bottom flask fitted with a mechanical stirrer and a Dean-Stark condenser. The reaction mixture was refluxed for 6-8 hours, while monitoring the reaction on TLC. After the completion of the reaction, the reaction mass was cooled to 10 °C and the solid thus obtained was filtered, washed with xylene (2 X 250 ml) and dried at 80 °C for 1-2 hours, to afford 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione (795g, Y=85%, P=98%).

**Example-10****One pot preparation of 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione**

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), toluene (2.5 L), p-toluene sulphonic acid (39 g, 0.21 M) and iodine (2 g, catalytic) were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Ethylbromoacetate (341 g, 2.05M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with toluene (2 X 500 ml). The combined organic layers were taken in a 5 L 4-neck

round bottom flask fitted with a mechanical stirrer and a Dean-Stark condenser. Thiazolidine-2,4-dione (239 g, 2.05 M), piperidine (30 ml, 0.30 M) and benzoic acid (30 g, 0.20 M) were added and the reaction mixture was refluxed for 6-8 hours, while monitoring the reaction on TLC. After the completion of the reaction, the reaction mass was cooled to 10 °C and the solid thus obtained was filtered, washed with toluene (2 X 250 ml) and dried at 80 °C for 1-2 hours to afford 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione (473 g, Y=75%, P=98%).

#### Example-11

##### Alternative one pot preparation of 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), xylene (2.5 L), p-toluenesulfonic acid (39 g, 0.21 M) and iodine (2 g, catalytic) were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Ethylbromoacetate (341 g, 2.05M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with toluene (2 X 500 ml). The combined organic layer was taken in a 5 L 4-neck round bottom flask fitted with a mechanical stirrer and a Dean-Stark condenser. Thiazolidine-2,4-dione (234 g, 2.00 M), piperidine (30 ml, 0.30 M) and benzoic acid (30 g, 0.20 M) were added and the reaction mixture was refluxed for 6-8 hours, while monitoring the reaction on TLC. After the completion of the reaction, the reaction mass was cooled to 10 °C and the solid thus obtained was filtered, washed with xylene (2 X 250 ml) and dried at 80 °C for 1-2 hours, to

afford 5-[4-[(carboethoxy) methoxy]benzylidene]thiazolidine-2,4-dione (474 g, Y=75%, P=98%).

#### **Example-12**

##### **Preparation of 5-[4-[(carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione**

4-((Carbomethoxy)methoxy)benzaldehyde obtained by following a procedure described in Example 6 or 7 (500 g, 2.58 M), thiazolidine-2,4-dione (302 g, 2.58 M), piperidine (38 ml, 0.46 M), benzoic acid (38 g, 0.31 M) and toluene (3 L) were taken in a 5 L 4-neck round bottom flask fitted with a mechanical stirrer and a Dean-Stark condenser. The reaction mixture was refluxed for 6-8 hours, while monitoring the reaction on TLC. After the completion of the reaction, the reaction mass was cooled to 10 °C and the solid thus obtained was filtered, washed with toluene (2 X 250 ml) and dried at 80 °C for 1-2 hours, to afford 5-[4-[(carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione (645g, Y=85%, P=98%).

#### **Example-13**

##### **Alternative preparation of 5-[4-[(carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione**

4-((Carbomethoxy)methoxy)benzaldehyde obtained by following a procedure described in Example 6 or 7 (500 g, 2.58 M), thiazolidine-2,4-dione (302 g, 2.58 M), piperidine (38 ml, 0.46 M), benzoic acid (38 g, 0.31 M) and xylene (3 L) were taken in a 5 L 4-neck round bottom flask fitted with a mechanical stirrer and a Dean-Stark condenser. The reaction mixture was refluxed for 6-8 hours, while

monitoring the reaction on TLC. After the completion of the reaction, the reaction mass was cooled to 10 °C and the solid thus obtained was filtered, washed with xylene (2 X 250 ml) and dried at 80 °C for 1-2 hours, to afford 5-[4-[(carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione (647 g, Y=85%, P=98%).

#### **Example-14**

##### **One pot preparation of 5-[4-[(carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione**

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), toluene (2.5 L), p-toluenesulfonic acid (39 g, 0.21 M) and iodine (2 g, catalytic) were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Methylchloroacetate (223 g, 2.05M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with 2 X 500 ml of toluene. The combined organic layers were taken in a 5 L 4 neck round bottom flask fitted with a mechanical stirrer and a Dean-Stark condenser. Thiazolidine-2,4-dione (234 g, 2.00 M), piperidine (30 ml, 0.30 M) and benzoic acid (30 g, 0.20 M) were added and the reaction mixture was refluxed for 6-8 hours, while monitoring the reaction on TLC. After the completion of the reaction, the reaction mass was cooled to 10 °C and the solid thus obtained was filtered, washed with toluene (2 X 250 ml) and dried at 80 °C for 1-2 hours to afford 5-[4-[(carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione (460 g, Y=76%, P=98%).

**Example-15****Alternative one pot preparation of 5-[4-[(carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione**

4-Hydroxybenzaldehyde (250 g, 2.05 M), potassium carbonate (565 g, 4.09 M), xylene (2.5 L), p-toluenesulfonic acid (39 g, 0.21 M) and iodine (2 g, catalytic) were taken in a 5 L 4-neck round bottom flask with mechanical stirrer and a Dean-Stark condenser. Methylchloroacetate (223 g, 2.05M) was added and the reaction was refluxed for 6-8 hours, under azeotropic removal of water, while monitoring the reaction on TLC. After the completion of the reaction, water was added and the organic layer separated while the aq. layer was extracted with 2 X 500 ml of toluene. The combined organic layers were taken in a 5 L 4-neck round bottom flask fitted with a mechanical stirrer and a Dean-Stark condenser. Thiazolidine-2,4-dione (234 g, 2.00 M), piperidine (30 ml, 0.30 M) and benzoic acid (30 g, 0.20 M) were added and the reaction mixture was refluxed for 6-8 hours, while monitoring the reaction on TLC. After the completion of the reaction, the reaction mass was cooled to 10 °C and the solid thus obtained was filtered, washed with xylene (2 X 250 ml) and dried at 80 °C for 1-2 hours to afford 5-[4-[(carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione (465 g, Y=77%, P=98%).

**Example-16****Preparation of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione**

In an autoclave vessel (2 L), Raney Ni (60 ml) was placed along with ethyl acetate (600 ml), after washing the Raney Ni consecutively with water (2 X 250 ml), methanol (2 X 150 ml) and ethyl acetate (2 X 100 ml). Then 5-[4-

[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 8-11 (100 g, 0.33 M), was charged to the vessel and ethyl acetate (600 ml) added. The mass was kept for hydrogenation at 400 psi hydrogen pressure, at room temperature for 20-30 h and the reaction was monitored on HPLC. After completion of the reaction, the catalyst was filtered and the filtrate was evaporated under reduced pressure to yield an oil. This oil was kept under high vacuum to afford the solid 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione in crude form (95-97 g; Yield : 94-96%; Purity : 86-95% (HPLC)). Crude 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione thus obtained was dissolved in hot methanol (200 ml) and transferred to a 2 L three neck round bottom flask fitted with a mechanical stirrer and liquid addition funnel. Demineralised water (400 ml) was added dropwise to the reaction mixture through the addition funnel, over a period of 30 min. with vigorous stirring during which a white compound precipitated out. Stirring was continued for further 30 min. and second portion of water (200 ml) was added while stirring, over a period of 15 min. to ensure complete precipitation of the product. Stirring was continued for further 1 h. The product was filtered, washed with water (200 ml) and dried under vacuum to yield pure 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione (85 g, Y=85%, P=99%).

#### **Example-17**

##### **Preparation of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione**

In an autoclave vessel 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 8-11 (100 g) and ethyl acetate (600 ml) were placed. Raney Ni (60 ml), pre washed with

water (2 X 100 ml), methanol (2 X 100 ml) and ethyl acetate (1 X 100 ml) consecutively, was transferred into the vessel with ethyl acetate (600 ml). Then the mass was kept for hydrogenation at 400 psi hydrogen pressure and room temperature for 24 h. The catalyst was filtered and the filtrate was evaporated and dried over vacuum to obtain 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione (98 g, Y=97.3 %, P= 92 %).

#### **Example 18**

##### **Preparation of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione**

In an autoclave vessel 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 8-11 (100 g) and ethyl acetate (600 ml) were placed. Raney Ni (60 ml), pre-washed with water (2 X 100 ml), methanol (2 X 100 ml) and ethyl acetate (1 X 100 ml) consecutively was transferred into the reaction vessel with ethyl acetate (600 ml). Then the mass was kept for hydrogenation at 200 psi hydrogen pressure and room temperature for 30 h. The catalyst was filtered and the filtrate was evaporated and dried over vacuum to obtain 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione (100 g, Y=97.35 %, P=80 %).

#### **Example 19**

##### **Preparation of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione**

In an autoclave vessel 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 8-11 (100 g) and ethyl acetate (600 ml) were placed. Raney Ni (60 ml) pre-washed with water (2 X 100 ml), methanol (2 X 100 ml), ethyl acetate (1 X 100 ml)



consecutively was transferred into the reaction vessel with ethyl acetate (600ml). Then the mass was kept for hydrogenation at 100 psi hydrogen pressure and room temperature for 70 h. The catalyst was filtered and the filtrate was evaporated and dried over vacuum to obtain 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione (100 g, Y= 99.35 %, P=78 %).

#### **Example 20**

##### **Preparation of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione**

In an autoclave vessel 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 8-11 (40 g) and ethyl acetate (400 ml) were placed. Raney Ni (32 ml) pre-washed with water (2 X 100 ml), methanol (2 X 100 ml), ethyl acetate (1 X 100 ml) was transferred with ethyl acetate (400 ml) into the vessel. The mass was kept for hydrogenation at 400 psi hydrogen pressure and 50 °C – 60 °C temperature for 11 h. The catalyst was filtered and the filtrate was evaporated and dried over vacuum to obtain 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione (37 g, Y=92 %, P=80.17 %).

#### **Example 21**

##### **Preparation of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione**

In a 3 L four necked round bottom flask with a mechanical stirrer, thermometer socket, condenser and a gas sparger, 5-[4-[(carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 8-11 (100 g), ethyl acetate (600 ml) were placed. Raney Ni (100 ml) pre-washed with water (2 X 150 ml), methanol

(2 X 150 ml), ethyl acetate (1 X 150 ml) was transferred with ethyl acetate (600 ml) into the vessel. Then hydrogen gas was bubbled into the solution at room temperature for 36 h. The catalyst was filtered and the filtrate was evaporated and dried over vacuum to obtain 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione (100 g, Y=99.35 %, P=73 %).

#### **Example-22**

##### **Alternative procedure for the preparation of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione**

5-[4-[(Carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 8-11 (100 g, 0.33 M), magnesium (95 g, 3.96 M) and methanol (50 ml) were taken into a 5 L round bottom flask fitted with a mechanical stirrer and stirred for 10 minutes at room temperature during which period the magnesium starts reacting, as evinced by effervescence. Ethanol was added and the temperature of the reaction mass was maintained at 20-25 °C for 12 hours, while monitoring the reaction by HPLC. After the complete reduction, the reaction mass was cooled to 5 °C and the pH was adjusted to 2, using conc. sulphuric acid and the reaction mixture was refluxed for a further 12 hours period, while monitoring the reaction by TLC. After complete esterification, the reaction mixture was cooled to room temperature and the magnesium salts (~500 g) were filtered. The filtrate was concentrated under vacuum and the residue was dissolved in ethyl acetate (250 ml) and pure 5-[4-[(carboethoxy)methoxy]benzyl] thiazolidine-2,4-dione was precipitated as a white solid by adding pet. ether (125 ml) followed by stirring at room temperature for 1 hour (59 g, Y=61%, P=97%).

**Example-23****Preparation of 5-[4-[(carbomethoxy)methoxy]benzyl]thiazolidine-2,4-dione**

5-[4-[(Carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 8-11 (100 g, 0.33 M), magnesium (95 g, 3.96 M) and methanol (2 L) were taken into a 5 L round bottom flask fitted with a mechanical stirrer and stirred for 10 minutes at room temperature during which period the magnesium starts reacting, as evinced by effervescence. The temperature of the reaction mass was maintained at 20-25 °C for 12 hours, while monitoring the reaction by HPLC. After the complete reduction and trans-esterification, the reaction mass was cooled to 5 °C and the pH was adjusted to 2 using conc. sulphuric acid and the reaction mixture was refluxed for a further 12 hours period, while monitoring the reaction by TLC. After complete esterification, the reaction mixture was cooled to room temperature and the magnesium salts (~500 g) were filtered. The filtrate was concentrated under vacuum and the residue was dissolved in ethyl acetate (250 ml) and pure 5-[4-[(carbomethoxy)methoxy]benzyl] thiazolidine-2,4-dione was precipitated as a white solid by adding pet. ether (125 ml) followed by stirring at room temperature for 1 hour (60 g, Y=62%, P=97%).

**Example-24****Alternative procedure for the preparation of 5-[4-[(carbomethoxy)methoxy]benzyl] thiazolidine-2,4-dione**

5-[4-[(Carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione, obtained by following a procedure described in any of Examples 12-15 (100 g, 0.34 M), magnesium (95 g, 3.96 M) and methanol (2 L) were taken into a 5 L round

bottom flask fitted with a mechanical stirrer and stirred for 10 minutes at room temperature during which period the magnesium starts reacting, as evinced by effervescence. The temperature of the reaction mass was maintained at 20-25 °C for 12 hours, while monitoring the reaction by HPLC. After the complete reduction and trans-esterification, the reaction mass was cooled to 5 °C and the pH was adjusted to 2 using conc. sulphuric acid and the reaction mixture was refluxed for a further 12 hours period, while monitoring the reaction by TLC. After complete esterification, the reaction mixture was cooled to room temperature and the magnesium salts (~500 g) were filtered. The filtrate was concentrated under vacuum and the residue was dissolved in ethyl acetate (250 ml) and pure 5-[4-[(carboethoxy)methoxy] benzyl]thiazolidine-2,4-dione was precipitated as a white solid by adding pet. ether (125 ml) followed by stirring at room temperature for 1 hour (60 g, Y=60%, P=97%).

#### **Example-25**

##### **Preparation of 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione**

A suspension of the 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 16-22 (135 g, 0.44 M) and water (540 ml, 4 times w/v) was taken in a round bottom flask fitted with a mechanical stirrer. Aq. sodium hydroxide solution (37 g of NaOH in 135 ml of water) was added slowly over a period of 5-10 minutes at 20-25 °C. Stirring was continued at ambient temperature for a further period of 2-3 h, while monitoring the reaction by TLC. After the completion of reaction, the pH of the reaction mixture was adjusted to 2 using conc. HCl (temp. raises to ~ 40-45 °C) and allowed to attain room temperature. The mass was cooled to ~ 10-15 °C and the solid thus obtained was filtered and dried at 60-70 °C under 1-2 mm Hg of

vacuum to afford 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione (121 g, Y=99%, P=99.2%).

#### **Example-26**

##### **Alternative preparation of 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione**

5-[4-[(Carboethoxy)methoxy]benzylidene]thiazolidine-2,4-dione, obtained by following a procedure described in any of Examples 8-11 (100 g, 0.33 M), magnesium (95 g, 3.96 M) and methanol (2 L) were taken into a 5 L round bottom flask fitted with a mechanical stirrer and stirred for 10 minutes at room temperature during which period the magnesium starts reacting, as evinced by effervescence. The temperature of the reaction mass was maintained at 20-25 °C for 12 hours, while monitoring the reaction by HPLC. After the complete reduction and trans-esterification, water (2 L) was added to the reaction mass and stirred at ambient temperature for a further 12 hours period, while monitoring the reaction by TLC. After complete hydrolysis, the reaction mixture was acidified to pH 2 and extracted with ethyl acetate (3 X 200 ml). The combined organic extract was concentrated under vacuum. Pure 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione was precipitated as a white solid by adding pet. ether (125 ml) followed by stirring at room temperature for 1 hour (60 g, Y=66%, P=97%).

#### **Example-27**

##### **Alternate procedure for the preparation of 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione**

A suspension of 5-[4-[(carbomethoxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 23-24 (135 g, 0.46 M) and water (540 ml, 4 times w/v) was taken in a round bottom flask fitted with a mechanical stirrer. Aq. sodium hydroxide solution (37 g of NaOH in 135 ml of water) was added slowly over a period of 5-10 minutes at 20-25 °C. Stirring was continued at ambient temperature for a period of 2-3 h, while monitoring the reaction by TLC. After the completion of the reaction, the pH of the reaction mixture was adjusted to 2 using conc. HCl (temp. raises to 40-45 °C) and allowed to attain room temperature. The mass was cooled to 10-15 °C and the solid thus obtained was filtered and dried at 60-70 °C under 1-2 mm Hg of vacuum to afford 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione (121 g, Y=99%, P=99.2%).

#### Example-28

##### Alternative Preparation of 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione

5-[4-[(Carbomethoxy)methoxy]benzylidene]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 12-15 (100 g, 0.34 M), magnesium (95 g, 3.96 M) and methanol (2 L) were taken into a 5 L round bottom flask fitted with a mechanical stirrer and stirred for 10 minutes at room temperature during which period the magnesium starts reacting, as evinced by effervescence. The temperature of the reaction mass was maintained at 20-25 °C for 12 hours, while monitoring the reaction by HPLC. After the complete reduction and trans-esterification, water (2 L) was added to the reaction mass and stirred at ambient temperature for a further 12 hours period, while monitoring the reaction by TLC. After complete hydrolysis, the reaction mixture was acidified to

pH 2 and extracted with ethyl acetate (3 X 200 ml). The combined organic extract was concentrated under vacuum. Pure 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione was precipitated as a white solid by adding pet. ether (125 ml) followed by stirring at room temperature for 1 hour (62 g, Y=65%, P=97%).

#### **Example-29**

##### **Preparation of 5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione**

A suspension of 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 25-28 (100 g, 0.356 M), N-methyl anthranilamide (58.7 g, 0.391 M), and p-toluenesulphonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 12-15 h while monitoring the reaction by TLC. After completion of the reaction, the reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a dropping funnel. The reaction mass was allowed to attain room temperature while stirring and the solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (100 g, Y=71%, P=98%).

#### **Example-30**

##### **Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione**

A suspension of 5-[4-[(carboxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 25-28 (100 g, 0.36 M), N-methyl anthranilamide (58.7 g, 0.39 M), xylene (100 ml) and p-toluenesulfonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 12-15 h while monitoring the reaction by TLC. After completion of the reaction, reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a dropping funnel. The reaction mass was allowed to attain room temperature while stirring and the solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (95 g, Y=68%, P=98%).

### Example-31

#### Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione

A suspension of 5-[4-[(carbomethoxy)methoxy]benzyl]thiazolidine-2,4-dione (100 g, 0.34 M) obtained in Example 23 or 24, N-methyl anthranilamide (58.7 g, 0.39 M) and p-toluenesulfonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 45-55 h while monitoring the reaction by TLC. After completion of the reaction, the reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a dropping funnel. The reaction mass was allowed to attain room temperature while stirring and the



solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (71 g, Y=53%, P=98%).

#### **Example-32**

##### **Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione**

A suspension of 5-[4-[(carbomethoxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in Example 23 or 24 (100 g, 0.34 M), N-methyl anthranilamide (58.7 g, 0.39 M), xylene (100 ml) and p-toluenesulfonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 45-55 h while monitoring the reaction by TLC. After completion of the reaction, the reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a dropping funnel. The reaction mass was allowed to attain room temperature while stirring and the solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (66 g, Y=49%, P=98%).

#### **Example-33**

##### **Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione**

A suspension of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 16-22 (100 g, 0.32 M), N-methyl anthranilamide (58.7 g, 0.39 M) and p-toluenesulfonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 45-55 h while monitoring the reaction by TLC. After completion of the reaction, the reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a dropping funnel. The reaction mass was allowed to attain room temperature while stirring and the solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (68 g, Y=53%, P=98%).

#### **Example-34**

##### **Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione**

A suspension of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 16-22 (100 g, 0.32 M), N-methyl anthranilamide (58.7 g, 0.39 M), xylene (100 ml) and p-toluenesulfonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 45-55 h while monitoring the reaction by TLC. After completion of the reaction, the reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a dropping funnel. The reaction mass was

allowed to attain room temperature while stirring and the solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (62 g, Y=48%, P=98%).

#### Example-35

##### Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione

A suspension of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 16-22 (100 g, 0.32 M), N-methyl anthranilamide (58.7 g, 0.39 M) and p-toluenesulfonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 10-15 h while monitoring the reaction by TLC. After completion of the reaction, the reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a dropping funnel. The reaction mass was allowed to attain room temperature while stirring and the solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (29 g, Y=23%, P=98%).

#### Example-36

##### Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione

A suspension of 5-[4-[(carboethoxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 16-22 (100 g, 0.32 M), N-methyl anthranilamide (58.7 g, 0.39 M), xylene (100 ml) and p-toluenesulfonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 10-15 h while monitoring the reaction by TLC. After completion of the reaction, the reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a dropping funnel. The reaction mass was allowed to attain room temperature while stirring and the solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (29 g, Y=23%, P=98%) .

#### **Example-37**

##### **Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione**

A suspension of 5-[4-[(carbomethoxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in Examples 23 or 24 (100 g, 0.34 M), N-methyl anthranilamide (58.7 g, 0.39 M) and p-toluenesulfonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 10-15 h while monitoring the reaction by TLC. After completion of the reaction, reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a

dropping funnel. The reaction mass was allowed to attain room temperature while stirring and the solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (28 g, Y=23%, P=98%).

#### Example-38

##### Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione

A suspension of 5-[4-[(carbomethoxy)methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in Examples 23 or 24 (100 g, 0.34 M), N-methyl anthranilamide (58.7 g, 0.39 M), xylene (100 ml) and p-toluenesulfonic acid (~200 mg) was taken in a round bottom flask fitted with a mechanical stirrer, oil bath and Dean-Stark condenser. The reaction mixture was heated to reflux (Internal temperature 150-155 °C, oil bath temperature 170-180 °C) for a period of 10-15 h while monitoring the reaction by TLC. After completion of the reaction, reaction mass was cooled to 80 °C and methanol (700 ml) was added slowly through a dropping funnel. The reaction mass was allowed to attain room temperature while stirring and the solid thus obtained was filtered and washed with methanol (150 ml) and dried at 100-120 °C for 1 h to afford 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione as a white solid (30 g, Y=23%, P=98%).

**Example-39****Preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione potassium salt**

5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4dione obtained by following a procedure described in any of Examples 29-38 (100 g, 0.25 M) was dissolved in 1 L of xylene : MeOH (1 : 1) mixture at 80-90 °C, treated with decolourising carbon (20 g) and filtered. To the filtrate was added potassium hydroxide solution (15.6 g of potassium hydroxide dissolved in 200 ml of methanol) slowly over a period of 5-10 min. at 60-70 °C. Stirring was continued at ambient temperature for a period of 1 h. The solid obtained was filtered, washed with methanol (300 ml) and dried at 120 °C for 1 h to yield 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione potassium salt as an off- white solid (98 g, Y=89%, P=99.5%).

✓ **Example-40****Alternative preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione potassium salt**

5-[4-[[3-Methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione obtained by following a procedure described in any of Examples 29-38 (100 g, 0.25 M) was dissolved in 1 L of xylene : MeOH (1 : 1) mixture at 80-90 °C, treated with decolourising carbon (20 g) and filtered. To the filtrate was added potassium t-butoxide solution (31.56 g of potassium t-butoxide dissolved in 200 ml of methanol) slowly over a period of 5-10 min. at 60-70 °C. Stirring was continued at ambient temperature for a period of 1 h. The solid obtained was filtered, washed with methanol (300 ml) and dried at 120 °C for 1 h to yield

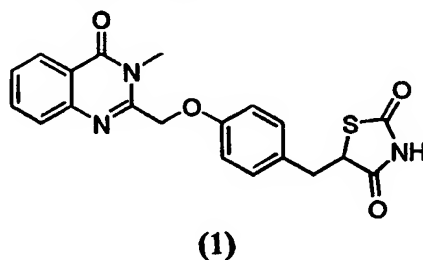
5-[4-[[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione potassium salt as a white solid (100 g, Y=91%, P=99.6%).

**Advantages of the invention:**

- The process is simple and economical.
- Multisolvent systems are replaced with single solvent system.
- Use of expensive Pd/C for reduction is replaced with relatively inexpensive reagents like Raney Nickel or Magnesium /alcohol having 1 to 4 carbon atoms.
- Activation of the acid prior to condensation with N-methyl anthranilamide is avoided. Direct condensation is easier to handle as the reaction is insensitive to moisture.
- Expensive and hazardous potassium t-butoxide for potassium salt formation is replaced with potassium hydroxide.

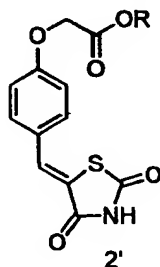
We claim

1. A process for the preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1)

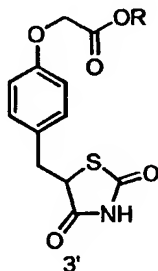


which comprises

- a) reducing the compound of the formula (2')



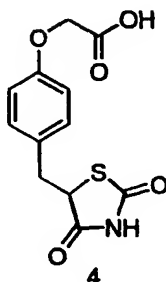
where R represents a (C<sub>1</sub>-C<sub>4</sub>)alkyl group and if desired reesterifying using sulphuric acid at a temperature in the range of 0 °C to 60 °C to obtain a compound of formula (3')



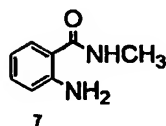
wherein R is as defined above,



- b) hydrolyzing the compound of formula (3') to obtain the acid of the formula (4),



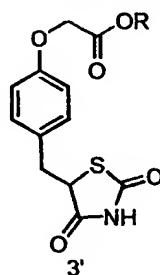
- c) condensing the acid of the formula (4) with N-methyl anthranilamide of the formula (7)



directly without any preactivation of the acid to produce the compound of formula (1) and if desired

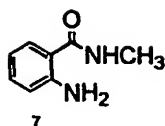
- d) converting the compound of formula (1) to pharmaceutically acceptable salts thereof.

2. A process as claimed in claim 1, wherein the compound of the formula (3')



where R is as defined above is condensed directly with N-methyl anthranilamide of the formula (7)

40



to obtain the compound of the formula (1).

3. A process as claimed in claims 1 and 2, wherein the reduction is carried out using Raney Nickel.
4. A process as claimed in claims 1 to 3, wherein the reduction is carried out for a period ranging from 8 to 70 hours, preferably ranging from 12 to 24 hours.
5. A process as claimed in claims 1 to 4, wherein the reduction is carried out at a pressure ranging from atmospheric pressure to 600 psi, preferably from atmospheric pressure to 400 psi.
6. A process as claimed in claims 1 to 5, wherein the reduction is carried out at a temperature in the range of 15 °C-70 °C preferably from 30 °C-60 °C.
7. A process as claimed in claims 1 to 6, wherein the purification of the crude reduced product is effected using lower alcohol containing 1-4 carbon atoms and followed by precipitation using water.
8. A process as claimed in claims 1 and 2, wherein the reduction in step (a) is carried out using Mg/alcohol having 1 to 4 carbon atoms.

9. A process as claimed in claims 1,2 and 8, wherein the reduction in step (a) is carried out using 4 to 12 equivalents of Mg, preferably 8 to 10 equivalents in alcohol having 1 to 4 carbon atoms.
10. A process as claimed in claims 1,2, 8 and 9, wherein the temperature of the reaction is in the range of 10 °C to 60 °C, preferably from 15 °C-30 °C.
11. A process as claimed in claims 1,2, 8-10, wherein the reaction time ranges from 2 to 15 hours, preferably ranging from 6 to 8 hours.
12. A process as claimed in claims 1,2, 8-11, wherein there is no effluent problem as all the magnesium salts are precipitated and filtered off.
13. A process as claimed in claim 1, wherein the reaction time in condensation step (c) ranges from 6 to 20 hours, preferably ranging from 10 to 12 hours.
14. A process as claimed in claim 2, wherein the reaction time ranges from 5 to 30 hours, preferably ranging from 6 to 20 hours.
15. A process as claimed in claims 1 and 2, wherein step (d) the pharmaceutically acceptable salt may be prepared by reacting the compound of formula (I) with methanolic potassium hydroxide, potassium carbonate or potassium t-butoxide in the presence of solvent to yield potassium salt of the compound of formula (1).

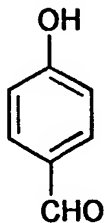
16. A process as claimed in claims 1,2 and 15, wherein the addition of methanolic potassium hydroxide, potassium carbonate or potassium t-butoxide is carried out at a temperature of 60-70 °C.

17. A process as claimed in claims 1,2, 15 and 16, wherein after the addition of potassium hydroxide, potassium carbonate or potassium t-butoxide, the reaction mixture is cooled to room temperature and maintaining it for 1 h at room temperature.

18. A process as claimed in claims 1, 2, 15-17, wherein solvent used is selected from xylene/methanol mixture in the ration of 1 : 1.

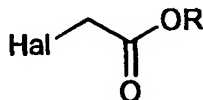
19. A process for the preparation of the compound of the formula (2'), which comprises

a). reacting p-hydroxybenzaldehyde of the formula (10)



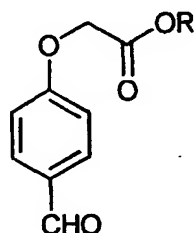
10

with alkylhaloacetate of the formula (11)



11

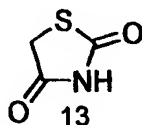
where Hal represents halogen atom like fluorine, chlorine or bromine and R is as defined earlier in the presence of aromatic hydrocarbon solvent, a base, alkyl or aryl sulphonic acid and iodine to obtain the compound of the formula (12)



12

and

b). condensing the compound of formula (12) where R is as defined earlier with thiazolidine-2,4-dione of the formula (13)



13

in the presence or absence of a solvent using catalysts to produce a compound of formula (2').

20. A process as claimed in claim 19, wherein the aromatic hydrocarbon solvent used is selected from benzene, toluene, xylene or mixtures thereof.

21. A process as claimed in claims 19-20, wherein the base used is selected from alkali and alkaline earth metal carbonates and bicarbonates such as potassium carbonate, potassium bicarbonate, sodium carbonate or calcium carbonate.

22. A process as claimed in claims 19-21, wherein the alkyl or aryl sulphonic acid used is selected from methane sulphonic acid, ethane sulphonic acid,

propane sulphonic acid, p-toluene sulphonic acid, benzene sulphonic acid or p-nitro benzene sulphonic.

23. A process as claimed in claims 19-22, wherein the reaction time for step (a) ranges from 3 to 10 hours, preferably 5 to 7 hours.

24. A process as claimed in claims 19-23, wherein the reaction time for step (b) ranges from 6 to 8 hours.

25. A process as claimed in claims 19-24, wherein the solvent used in step (b) is selected from toluene or xylene.

26. A process as claimed in claims 19-25, wherein the catalysts used in step (b) are selected from benzoic acid and piperidine.

27. A process for the preparation of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1) substantially as herein described with reference to examples 29-38.

28. A process for the preparation of the compound of the formula (2') substantially as herein described with reference to examples 8-15.

29. A process for the preparation of potassium salt of 5-[4-[[3-methyl-4-oxo-3,4-dihydroquinazolin-2-yl]methoxy]benzyl]thiazolidine-2,4-dione of the formula (1) substantially as herein described with reference to examples 39-40.

# INTERNATIONAL SEARCH REPORT

Intern al Application No

PCT/IB 99/01530

**A. CLASSIFICATION OF SUBJECT MATTER**  
IPC 7 C07D417/12 C07D277/24

According to International Patent Classification (IPC) or to both national classification and IPC

**B. FIELDS SEARCHED**

Minimum documentation searched (classification system followed by classification symbols)  
IPC 7 C07D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

**C. DOCUMENTS CONSIDERED TO BE RELEVANT**

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No
X	WO 97 41097 A (REDDY S RESEARCH FOUNDATION) 6 November 1997 (1997-11-06) cited in the application page 57, preparation 37B page 29, line 20 - page 31, line 8; claims	1-18
X	EP 0 306 228 A (BEECHAM GROUP PLC) 8 March 1989 (1989-03-08) page 8, line 41 - line 48; claims	1,8-14
X	EP 0 454 501 A (SANKYO CO) 30 October 1991 (1991-10-30) page 25, line 31 - page 26, line 35 -/-	1,8-14

☒ Further documents are listed in the continuation of box C

☒ Patent family members are listed in annex

\* Special categories of cited documents :

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other means
- "P" document published prior to the international filing date but later than the priority date claimed

- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- "Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
- "&" document member of the same patent family

Date of the actual completion of the international search

3 December 1999

Date of mailing of the international search report

16/12/1999

Name and mailing address of the ISA  
European Patent Office, P B 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel (+31-70) 340-2040, Tx 31651 epo nl,  
Fax: (+31-70) 340-3016

Authorized officer

Henry, J

# INTERNATIONAL SEARCH REPORT

Intern: al Application No

PCT/IB 99/01530

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No
X	DALE N. ROBERTSON: "Ethyl 4-formylphenoxyacetate" JOURNAL OF ORGANIC CHEMISTRY., vol. 21, 1956, page 1190 XP002124679 AMERICAN CHEMICAL SOCIETY. EASTON., US ISSN: 0022-3263 the whole document	19,21
X	EP 0 787 727 A (SS PHARMACEUTICAL CO) 6 August 1997 (1997-08-06) pages 4,5,see second step;see pages 11,12 preparatio example5	19,24-28



# INTERNATIONAL SEARCH REPORT

Information on patent family members

Intern. Application No

PCT/IB 99/01530

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9741097 A	06-11-1997	AU 3719897 A	19-11-1997
		CA 2258949 A	06-11-1997
		CZ 9803850 A	14-04-1999
		EP 0958296 A	24-11-1999
		NO 986055 A	22-12-1998
EP 0306228 A	08-03-1989	AU 2173888 A	09-03-1989
		CA 1328452 A	12-04-1994
		CA 1339902 A	09-06-1998
		DK 490288 A	05-03-1989
		EP 0842925 A	20-05-1998
		JP 10194970 A	28-07-1998
		JP 10194971 A	28-07-1998
		JP 1131169 A	24-05-1989
		JP 2614497 B	28-05-1997
		JP 2817840 B	30-10-1998
		JP 9183771 A	15-07-1997
		JP 2837139 B	14-12-1998
		JP 9183726 A	15-07-1997
		JP 9183772 A	15-07-1997
		NZ 226027 A	26-03-1992
		PT 88410 A, B	31-07-1989
		SG 59988 A	22-02-1999
		US 5646169 A	08-07-1997
		US 5002953 A	26-03-1991
		US 5521201 A	28-05-1996
		US 5232925 A	03-08-1993
		US 5194443 A	16-03-1993
		US 5756525 A	26-05-1998
		US 5260445 A	09-11-1993
EP 0454501 A	30-10-1991	EP 0839812 A	06-05-1998
		JP 4225978 A	14-08-1992
		US 5387596 A	07-02-1995
EP 0787727 A	06-08-1997	CA 2196400 A	01-08-1997
		CN 1167764 A	17-12-1997
		JP 9268189 A	14-10-1997
		US 5710152 A	20-01-1998